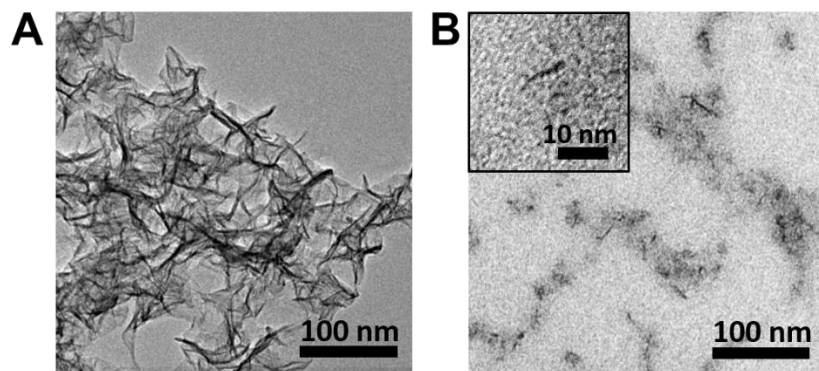


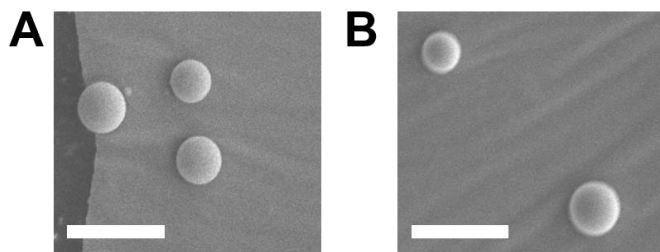
## Supporting Information

### Water-dispersible MnO<sub>2</sub> nanosheets complexed with alginate



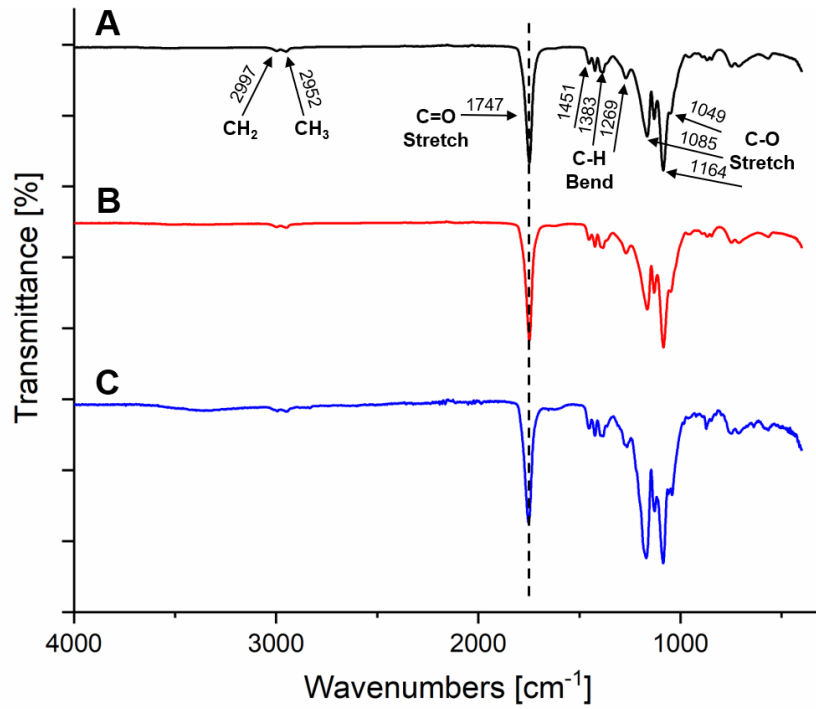
**Figure S1.** TEM images of (A) alginate-free MnO<sub>2</sub> nanosheets and (B) MnO<sub>2</sub> nanosheets complexed with alginate, denoted as MnO<sub>2</sub> nanosheets.

### SEM image of PLGA particles



**Figure S2.** SEM images of (A) PLGA particles and (B) PLGA /MnO<sub>2</sub> particles. Scale bar represents 5 μm.

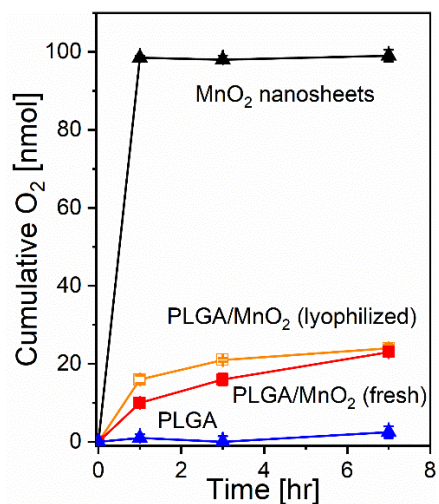
### FTIR spectra of PLGA particle



**Figure S3.** FTIR spectra of (A) Blank PLGA (B) PLGA/MnO<sub>2</sub> (C) PLGA/MnO<sub>2</sub>/thrombin particles.

### Cumulative O<sub>2</sub> generated through the H<sub>2</sub>O<sub>2</sub> decomposition

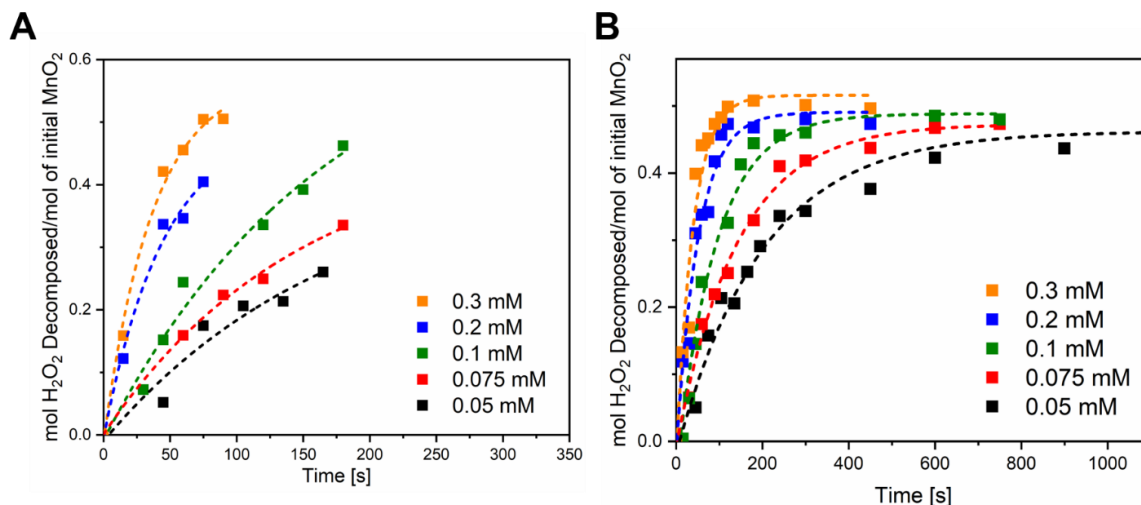
$$n_{O_2} = \frac{1}{2} \times (\text{initial } H_2O_2 \text{ concentration} - \text{final } H_2O_2 \text{ concentration}) \times H_2O_2 \text{ volume}$$



**Figure S4.** Cumulative O<sub>2</sub> generated through the H<sub>2</sub>O<sub>2</sub> decomposition. The cumulative O<sub>2</sub> was quantified by calculating the amount of H<sub>2</sub>O<sub>2</sub> that was decomposed. Data points and error bars represent the average and standard deviation of three different samples per condition, respectively.

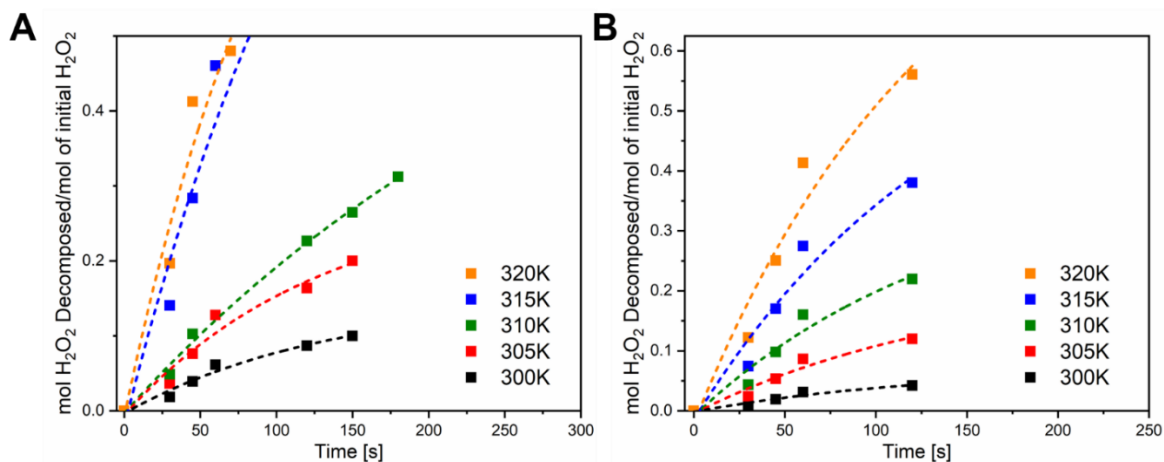
## H<sub>2</sub>O<sub>2</sub> decomposition curve under different conditions

### (a) Varying initial H<sub>2</sub>O<sub>2</sub> concentration



**Figure S5.** H<sub>2</sub>O<sub>2</sub> decomposition curves attained by adding (A) MnO<sub>2</sub> nanosheets and (B) lyophilized PLGA/MnO<sub>2</sub> particles into media with varied H<sub>2</sub>O<sub>2</sub> concentrations noted in each figure. Dash lines are for visual guide.

### (b) Varying reaction temperature



**Figure S6.** H<sub>2</sub>O<sub>2</sub> decomposition curves attained by adding (A) MnO<sub>2</sub> nanosheets and (B) lyophilized PLGA/MnO<sub>2</sub> particles into H<sub>2</sub>O<sub>2</sub>-containing media under different incubation temperatures noted in each figure. The initial H<sub>2</sub>O<sub>2</sub> concentration was kept constant at 0.2 mM. Dash lines are for visual guide.

**Theoretical calculation of the difference between internal pressure and external pressure of PLGA/MnO<sub>2</sub> particles**

Radius of the PLGA shell:  $r = 1.6 \mu\text{m}$

Radius of the PLGA core:  $r = 1.5 \mu\text{m}$

Density of PLGA polymer:  $\rho = 1.25 \text{ g/ml}$

Mass per PLGA shell:  $M_{PLGA} = \rho \times \frac{4}{3}\pi[(1.6 \mu\text{m})^3 - (1.5 \mu\text{m})^3] = 3.77 \times 10^{-12} \text{ g per particle}$

Number ( $N$ ) of particles per 10 mg of PLGA polymer:  $N_{\text{particle}} = 10 \text{ mg}/M_{PLGA} = 2.65 \times 10^9$

Total volume of 10 mg PLGA particle:  $V = N_{\text{particle}} \times \frac{4}{3}\pi(1.6 \mu\text{m})^3 = 45.44 \mu\text{L}$

Mole of O<sub>2</sub> gas generated inside the PLGA particle immersed in 100 $\mu\text{M}$  H<sub>2</sub>O<sub>2</sub> solution:

$$n_{O_2} = \frac{1}{2} \times (\text{initial } H_2O_2 \text{ concentration} - \text{final } H_2O_2 \text{ concentration}) \times H_2O_2 \text{ volume}$$

Pressure change due to O<sub>2</sub> generation:

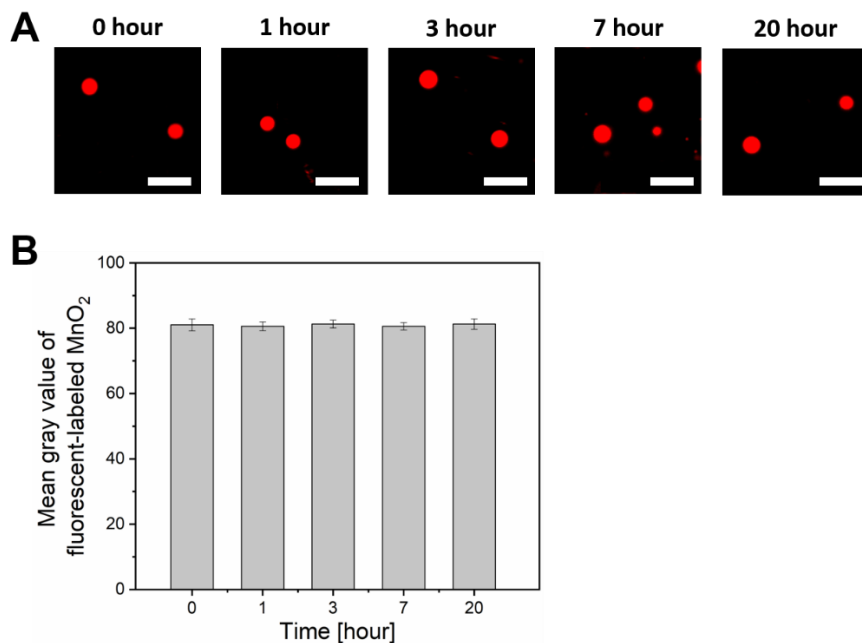
$$P_{O_2} V = n_{O_2} RT = P_{O_2} \times 45.44 \mu\text{L} = n_{O_2} \times 0.08206 \text{ atm} \cdot \text{Lmol}^{-1}\text{K}^{-1} \times 310 \text{ K}$$

$$\Delta P = \Delta P_{O_2} + \Delta P_{H_2O} \approx \Delta P_{O_2} = P_{O_2} - 0 = P_{O_2}$$

Group	Initial H <sub>2</sub> O <sub>2</sub> concentration ( $\mu\text{M}$ )	Final H <sub>2</sub> O <sub>2</sub> concentration ( $\mu\text{M}$ )	Generated O <sub>2</sub> (nmol)	$\Delta P$ (mmHg)
Lyophilized PLGA/MnO <sub>2</sub>	200	152	24	10.3
Fresh PLGA/MnO <sub>2</sub>	200	154	23	9.8
PLGA w/o MnO <sub>2</sub>	200	195	2.5	2.1

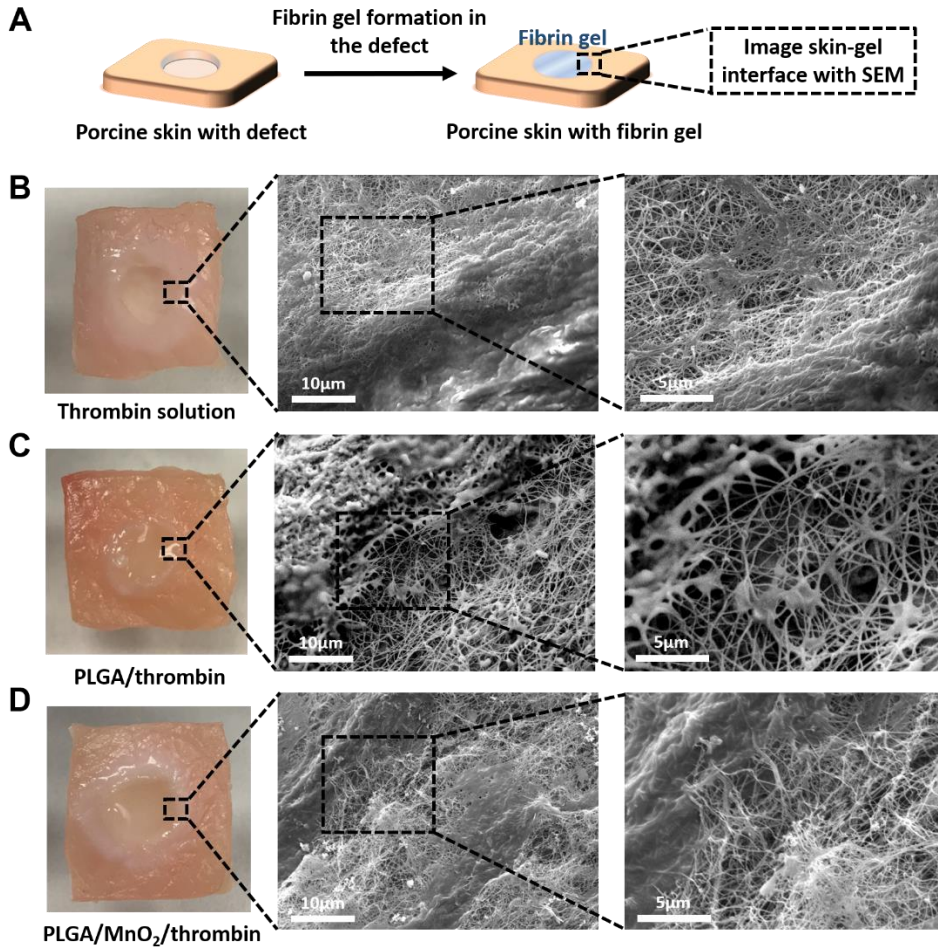
**Table S1.** O<sub>2</sub> generation and the pressure change during H<sub>2</sub>O<sub>2</sub> decomposition.

## Quantitative analysis of fluorescent-labeled $\text{MnO}_2$ intensity change over time



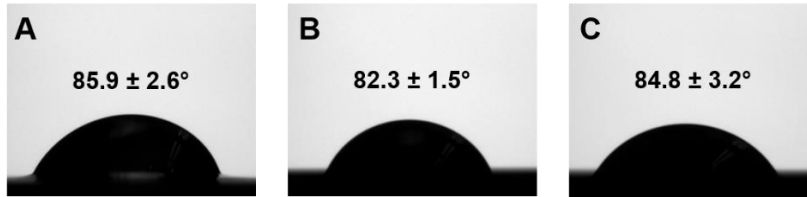
**Figure S7.** (A) Confocal images of PLGA/ $\text{MnO}_2$  microparticles incubated with  $\text{H}_2\text{O}_2$  at different time points. Alginate associating with  $\text{MnO}_2$  nanosheets were labeled with red-colored rhodamine. Scale bar represents 5  $\mu\text{m}$ . (B) Mean gray value of fluorescence from the PLGA/ $\text{MnO}_2$  microparticles with  $\text{H}_2\text{O}_2$  at different time points. Data points and error bars represent the average and standard deviation of three different samples per condition, respectively.

## Evaluation of the bio-adhesion of fibrin gel



**Figure S8.** (A) Schematic illustration of the experimental set-up used to evaluate the bio-adhesion of the fibrin gel to a porcine skin explant. The porcine skin explant was punched to simulate wounds where fibrin gels are implanted. The punched wounds were filled with fibrin gels (B-D) SEM images of the interface between skin and fibrin gel. The gel filled a hole created by punching out a center part of the square-shaped porcine skin. (B) Fibrin gel formed by mixing fibrinogen and thrombin solution. (C) Fibrin gel formed by mixing fibrinogen and PLGA/thrombin particles. (D) Fibrin gel formed by mixing fibrinogen and PLGA/MnO<sub>2</sub>/thrombin particles.

## Evaluation of surface water contact angle of fibrin gel



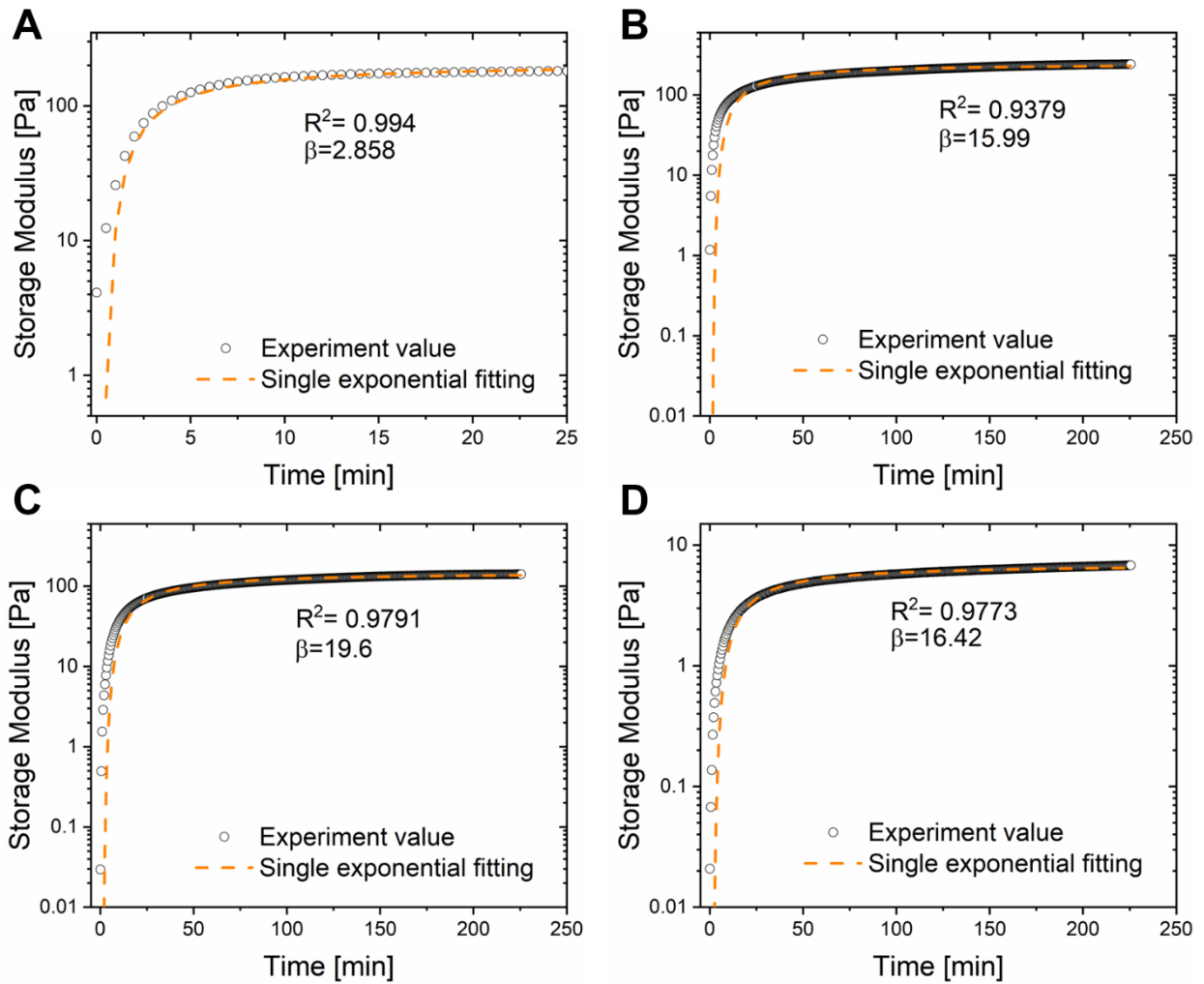
**Figure S9.** Contact angle measurements were taken with a goniometer using water as the test liquid for gel formed by mixing fibrinogen with (A) thrombin, (B) PLGA/thrombin particles, and (C) PLGA/MnO<sub>2</sub>/thrombin particles.



## Gelation curve fitting

### (1) Single exponential fitting

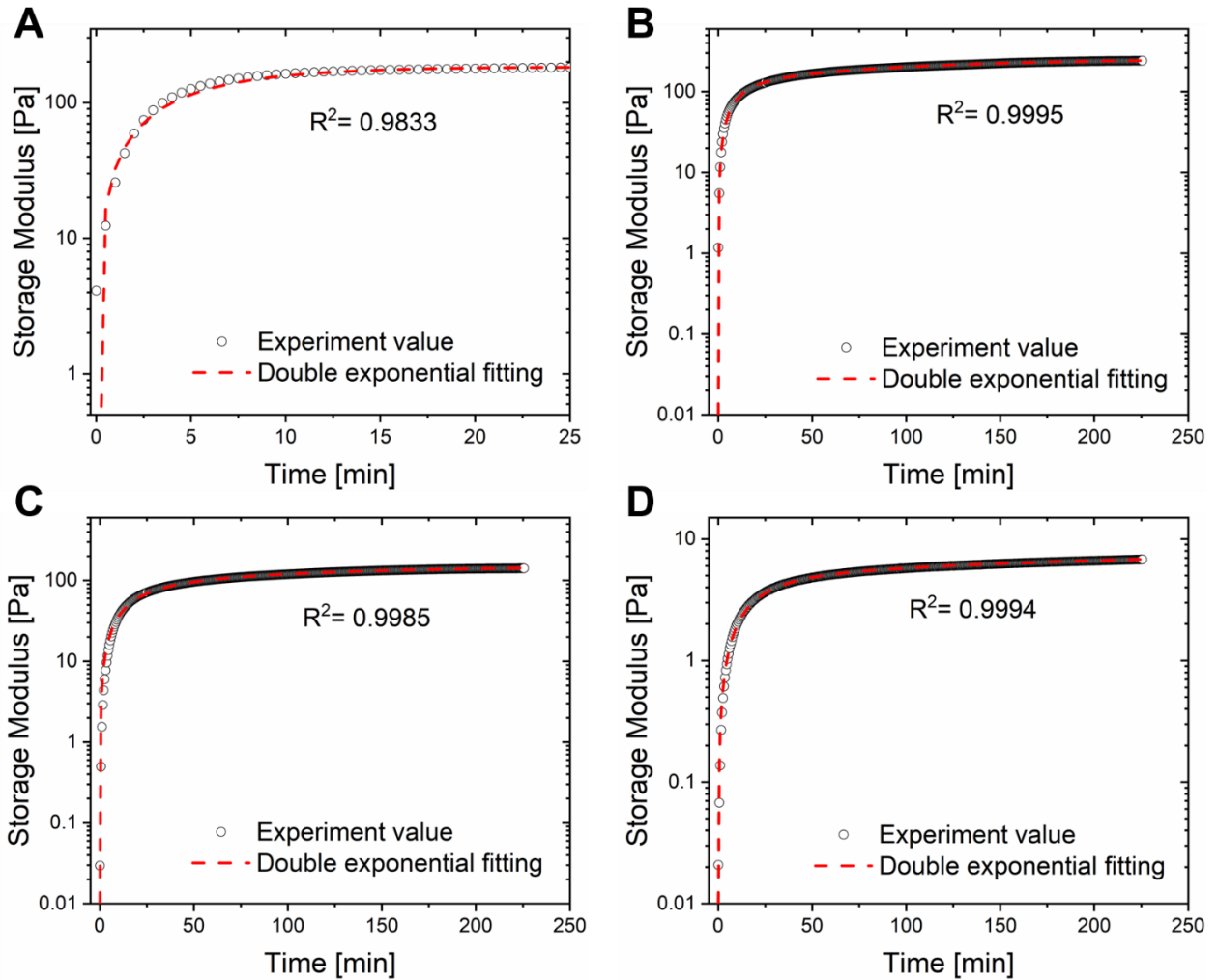
$G'(t) = G'_{eq} \exp(-\beta/t)$  where  $G'(t)$  is the storage modulus,  $G'_{eq}$  is the equilibrium storage modulus,  $\beta$  is the characteristic time, and  $t$  is time.



**Figure S10.** Single exponential fitting of the gelation curve. (A) Gelation curve attained by mixing thrombin solution with fibrinogen solution. (B) Gelation curve attained by mixing PLGA/MnO<sub>2</sub>/thrombin particles with a fibrinogen solution. (C) Gelation curve attained by mixing PLGA/thrombin particles with a fibrinogen solution. (D) Gelation curve of the mixture of blank PLGA particles, thrombin solution, and fibrinogen solution.

## (2) Double exponential fitting

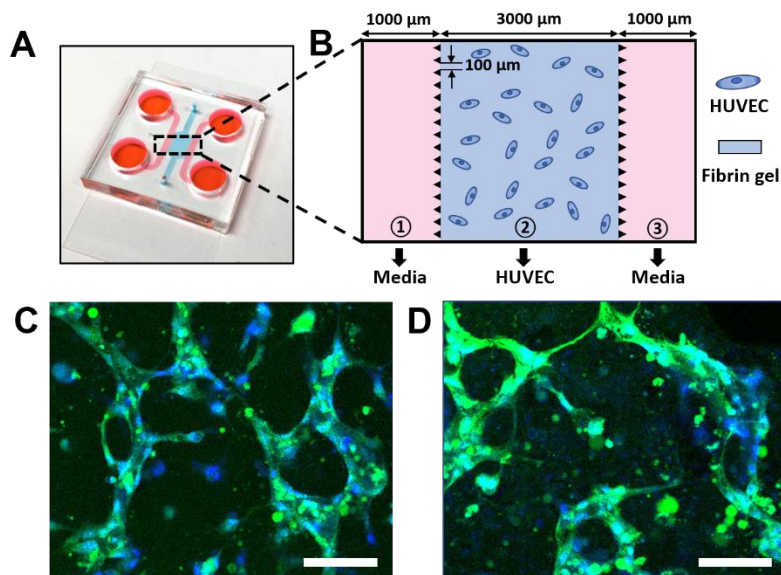
$G'(t) = G'_0 + G'_1(1 - \exp(-\frac{t}{\tau_1})) + G'_2(1 - \exp(-\frac{t}{\tau_2}))$  where  $G'(t)$  is the storage modulus;  $G'_0$  is the storage modulus of the pre-gelled solutions;  $G'_1$  is the storage modulus contributed by the first branching and crosslinking reaction phase;  $G'_2$  is the storage modulus caused by the second lateral growth of fibrin polymers;  $t$  is the time; and  $1/\tau_1$  and  $1/\tau_2$  are the characteristic rates for branching and lateral growth, respectively.



**Figure S11.** Double exponential fitting of the gelation curve. (A) Gelation curve attained by mixing thrombin solution with fibrinogen solution. (B) Gelation curve attained by mixing

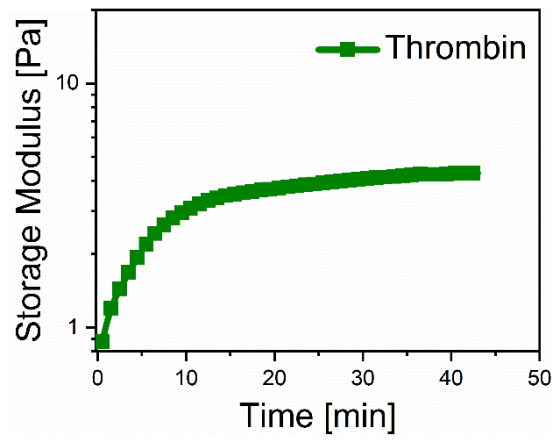
PLGA/MnO<sub>2</sub>/thrombin particles with a fibrinogen solution. (C) Gelation curve attained by mixing PLGA/thrombin particles with a fibrinogen solution. (D) Gelation curve of the mixture of blank PLGA particles, thrombin solution, and fibrinogen solution.

### In vitro angiogenesis assay



**Figure S12.** In vitro angiogenesis assay. (A) Schematic illustration of a microfluidic chip device used for the angiogenesis study in vitro. (B) The central portion features three channels. The fibrin gel encapsulating HUVECs filled the center channel (2); cell culture medium filled channels (1) and (3). Confocal laser scanning microscopic images of tube formation by human umbilical vein endothelial cells seeded in fibrin gel formed by mixing fibrinogen with (C) thrombin solution and (D) PLGA/MnO<sub>2</sub>/thrombin particles. Cells were immunostained with CD31 (in green). Cell nuclei were stained with DAPI (in blue). Scale bar represents 100 μm.

### Blood clot formation triggered by thrombin



**Figure S13.** Change in the storage modulus of the blood mixed with a thrombin solution.